

## AP EAPCET 2025 May 26 Shift 2 Question Paper

<b>Time Allowed :3 Hours</b>	<b>Maximum Marks :160</b>	<b>Total questions :160</b>
------------------------------	---------------------------	-----------------------------

### General Instructions

**Read the following instructions very carefully and strictly follow them:**

1. Duration of Exam: 3 Hours
2. Total Number of Questions: 160 Questions
3. Section-wise Distribution of Questions:
  - Physics - 40 Questions
  - Chemistry - 40 Questions
  - Mathematics - 80 Questions
4. Type of Questions: Multiple Choice Questions (Objective)
5. Marking Scheme: One mark awarded for each correct response
6. Negative Marking: There is no provision for negative marking.

**1. If the roots of the quadratic equation  $2x^2 - 4x + k = 0$  are real and equal, find the value of  $k$ .**

- (A) 2
  - (B) 3
  - (C) 4
  - (D) 5
- 

**2. If  $\tan \theta = \frac{4}{3}$ , find the value of  $\sin \theta$  in the first quadrant.**

- (A)  $\frac{3}{5}$
  - (B)  $\frac{4}{5}$
  - (C)  $\frac{3}{2}$
  - (D)  $\frac{4}{5}$
- 

**3. Find the midpoint of the line segment joining the points (3, 4) and (7, -2).**

- (A) (5, 1)
  - (B) (4, 1)
  - (C) (5, 2)
  - (D) (4, 2)
- 

**4. A bag contains 5 red and 3 blue balls. If one ball is drawn at random, what is the probability that it is red?**

- (A)  $\frac{3}{8}$
  - (B)  $\frac{5}{8}$
  - (C)  $\frac{3}{5}$
  - (D)  $\frac{2}{5}$
-

**5. If the roots of the quadratic equation  $ax^2 + bx + c = 0$  are in the ratio 3:4, then which of the following relationships holds between the coefficients?**

- (1)  $7b^2 = 48ac$
  - (2)  $48b^2 = 7ac$
  - (3)  $b^2 = 12ac$
  - (4)  $12b^2 = ac$
- 

**6. Find the derivative of the function  $f(x) = \sin^{-1}(2x\sqrt{1-x^2})$  with respect to  $x$ , for  $x \in (-\frac{1}{\sqrt{2}}, \frac{1}{\sqrt{2}})$ .**

- (1)  $\frac{2}{\sqrt{1-4x^2}}$
  - (2)  $\frac{1}{\sqrt{1-x^2}}$
  - (3)  $\frac{2}{\sqrt{1-x^2}}$
  - (4)  $\frac{1}{\sqrt{1-4x^2}}$
- 

**7. If  $\sin \theta + \cos \theta = \sqrt{2} \cos \theta$ , then the value of  $\tan \theta$  is**

- (1)  $\sqrt{2} + 1$
  - (2)  $\sqrt{2} - 1$
  - (3)  $1 - \sqrt{2}$
  - (4)  $\frac{1}{\sqrt{2}+1}$
- 

**8. The equation of the circle passing through the points (1, 1), (2, 2), and (3, 1) is**

- (1)  $x^2 + y^2 - 2x - 2y + 1 = 0$
  - (2)  $x^2 + y^2 - 4x - 2y + 5 = 0$
  - (3)  $x^2 + y^2 - 4x - 4y + 7 = 0$
  - (4)  $x^2 + y^2 - 2x - 4y + 3 = 0$
-

**9. A particle is projected at an angle of  $60^\circ$  to the horizontal with an initial speed of 20 m/s. If  $g = 10 \text{ m/s}^2$ , what is the maximum height reached by the particle?**

- (1) 15 m
  - (2) 20 m
  - (3) 30 m
  - (4) 45 m
- 

**10. Two capacitors of capacitances  $4 \mu\text{F}$  and  $6 \mu\text{F}$  are connected in series across a potential difference of 300 V. What is the potential difference across the  $4 \mu\text{F}$  capacitor?**

- (1) 100 V
  - (2) 120 V
  - (3) 180 V
  - (4) 200 V
- 

**11. A light ray passes from air ( $\mu = 1$ ) into a glass slab ( $\mu = 1.5$ ) at an angle of incidence of  $30^\circ$ . What is the angle of refraction inside the glass slab?**

- (1)  $19.5^\circ$
  - (2)  $22.5^\circ$
  - (3)  $35.0^\circ$
  - (4)  $45.0^\circ$
- 

**12. The rate constant for a first-order reaction is  $0.0693 \text{ min}^{-1}$ . What is the half-life of the reaction?**

- (1) 10 min
- (2) 15 min
- (3) 5 min
- (4) 20 min

---

**13. What is the IUPAC name of the compound  $\text{CH}_3\text{CH}_2\text{COCH}_2\text{CH}_3$ ?**

- (1) Pentan-3-one
- (2) Butan-2-one
- (3) Pentan-2-one
- (4) Hexan-3-one

---

**14. Which of the following elements has the highest first ionization energy?**

- (1) Sodium
- (2) Magnesium
- (3) Aluminum
- (4) Silicon

---

**15. The enthalpy change ( $\Delta H$ ) for the reaction  $\text{N}_2(g) + 3\text{H}_2(g) \rightarrow 2\text{NH}_3(g)$  is  $-92 \text{ kJ/mol}$ . What is the enthalpy change for the decomposition of 1 mol of  $\text{NH}_3(g)$  into its elements?**

- (1)  $+46 \text{ kJ/mol}$
  - (2)  $-46 \text{ kJ/mol}$
  - (3)  $+92 \text{ kJ/mol}$
  - (4)  $-92 \text{ kJ/mol}$
-