

# JEE Main 2024 Jan 29 (Shift 1) Question Paper

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1. If in a G.P. of 64 terms, the sum of all the terms is 7 times the sum of the odd terms of the G.P., then the common ratio of the G.P. is equal to:

- (1) 7
- (2) 4
- (3) 5
- (4) 6

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2. In an A.P., the sixth term  $a_6 = 2$ . If the  $a_1 a_4 a_5$  is the greatest, then the common difference of the A.P. is equal to:

- (1)  $\frac{3}{2}$
- (2)  $\frac{8}{5}$
- (3)  $\frac{2}{3}$
- (4)  $\frac{5}{8}$

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3. Given functions

$$f(x) = \begin{cases} 2 + 2x, & -1 \leq x < 0 \\ \frac{1-x}{3}, & 0 \leq x \leq 3 \end{cases}$$

and

$$g(x) = \begin{cases} -x, & -3 \leq x \leq 0 \\ x, & 0 < x \leq 1 \end{cases}$$

find the range of  $(f \circ g)(x)$ .

- (1)  $(0, 1]$
- (2)  $[0, 3)$

(3)  $[0, 1]$

(4)  $[0, 1)$

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**4. A fair die is thrown until the number 2 appears. What is the probability that 2 appears in an even number of throws?**

(1)  $\frac{5}{6}$

(2)  $\frac{1}{6}$

(3)  $\frac{5}{11}$

(4)  $\frac{6}{11}$

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**5. If  $z = \frac{1}{2} - 2i$ , is such that  $|z + 1| = \alpha z + \beta(1 + i)$ ,  $i = \sqrt{-1}$  and  $\alpha, \beta \in \mathbb{R}$ , then  $\alpha + \beta$  is equal to:**

(1)  $-4$

(2)  $3$

(3)  $2$

(4)  $-1$

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**6. Evaluate the limit**

$$\lim_{x \rightarrow \frac{\pi}{2}} \left( \frac{1}{\left(x - \frac{\pi}{2}\right)^3} \int_{\frac{\pi}{2}}^x \cos\left(\frac{1}{t^3}\right) dt \right)$$

**which is equal to:**

(1)  $\frac{3\pi}{8}$

(2)  $\frac{3\pi^2}{4}$

(3)  $\frac{3\pi^2}{8}$

(4)  $\frac{3\pi}{4}$

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**7. In a  $\triangle ABC$ , suppose  $y = x$  is the equation of the bisector of the angle  $B$  and the equation of the side  $AC$  is  $2x - y = 2$ . If  $2AB = BC$  and the points  $A$  and  $B$  are respectively  $(4, 6)$  and  $(\alpha, \beta)$ , then  $\alpha + 2\beta$  is equal to:**

(1)  $42$

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- (2) 39  
 (3) 48  
 (4) 45

**8. Let  $\vec{a}, \vec{b}$  and  $\vec{c}$  be three non-zero vectors such that  $\vec{b}$  and  $\vec{c}$  are non-collinear. If  $\vec{a} + 5\vec{b}$  is collinear with  $\vec{c}$ ,  $\vec{b} + 6\vec{c}$  is collinear with  $\vec{a}$ , and  $\vec{a} + \alpha\vec{b} + \beta\vec{c} = \vec{0}$ , then  $\alpha + \beta$  is equal to:**

- (1) 35  
 (2) 30  
 (3) -30  
 (4) -25

**9. Let  $(5, \frac{a}{4})$  be the circumcenter of a triangle with vertices  $A(a, -2)$ ,  $B(a, 6)$ , and  $C(\frac{a}{4}, -2)$ . Let  $\alpha$  denote the circumradius,  $\beta$  denote the area, and  $\gamma$  denote the perimeter of the triangle. Then  $\alpha + \beta + \gamma$  is:**

- (1) 60  
 (2) 53  
 (3) 62  
 (4) 30

**10. For  $x \in (-\frac{\pi}{2}, \frac{\pi}{2})$ , if**

$$y(x) = \int \frac{\csc x + \sin x}{\csc x \sec x + \tan x \sin^2 x} dx$$

**and**

$$\lim_{x \rightarrow -\frac{\pi}{2}} y(x) = 0,$$

**then  $y(\frac{\pi}{4})$  is equal to:**

- (1)  $\tan^{-1}\left(\frac{1}{\sqrt{2}}\right)$   
 (2)  $\frac{1}{2} \tan^{-1}\left(\frac{1}{\sqrt{2}}\right)$   
 (3)  $\frac{1}{\sqrt{2}} \tan^{-1}\left(\frac{1}{\sqrt{2}}\right)$   
 (4)  $\frac{1}{\sqrt{2}} \tan^{-1}\left(-\frac{1}{2}\right)$

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**11. If  $\alpha$ , with  $-\frac{\pi}{2} < \alpha < \frac{\pi}{2}$ , is the solution of  $4 \cos \theta + 5 \sin \theta = 1$ , then the value of  $\tan \alpha$  is:**

- (1)  $\frac{10-\sqrt{10}}{6}$
  - (2)  $\frac{10-\sqrt{10}}{12}$
  - (3)  $\frac{\sqrt{10}-10}{12}$
  - (4)  $\frac{\sqrt{10}-10}{6}$
- 

**12. A function  $y = f(x)$  satisfies**

$$f(x) \sin 2x + \sin x - (1 + \cos^2 x)f'(x) = 0$$

**with the condition  $f(0) = 0$ . Then  $f\left(\frac{\pi}{2}\right)$  is equal to:**

- (1) 1
  - (2) 0
  - (3) -1
  - (4) 2
- 

**13. Let  $O$  be the origin and the position vectors of  $A$  and  $B$  be  $2\hat{i} + 2\hat{j} + \hat{k}$  and  $2\hat{i} + 4\hat{j} + 4\hat{k}$  respectively. If the internal bisector of  $\angle AOB$  meets the line  $AB$  at  $C$ , then the length of  $OC$  is:**

- (1)  $\frac{2}{3}\sqrt{31}$
  - (2)  $\frac{2}{3}\sqrt{34}$
  - (3)  $\frac{3}{4}\sqrt{34}$
  - (4)  $\frac{3}{2}\sqrt{31}$
- 

**14. Consider the function  $f : \left[\frac{1}{2}, 1\right] \rightarrow \mathbb{R}$  defined by**

$$f(x) = 4\sqrt{2}x^3 - 3\sqrt{2}x - 1$$

**Consider the statements:**

- (I) The curve  $y = f(x)$  intersects the x-axis exactly at one point.
- (II) The curve  $y = f(x)$  intersects the x-axis at  $x = \cos \frac{\pi}{12}$ .

Then

- (1) Only (II) is correct
  - (2) Both (I) and (II) are incorrect
  - (3) Only (I) is correct
  - (4) Both (I) and (II) are correct
- 

**15. Let**

$$A = \begin{bmatrix} 1 & 0 & 0 \\ 0 & \alpha & \beta \\ 0 & \beta & \alpha \end{bmatrix}$$

**and  $|2A|^3 = 2^{21}$  where  $\alpha, \beta \in \mathbb{Z}$ . Then a value of  $\alpha$  is:**

- (1) 3
  - (2) 5
  - (3) 17
  - (4) 9
- 

**16. Let  $PQR$  be a triangle with  $R(-1, 4, 2)$ . Suppose  $M(2, 1, 2)$  is the midpoint of  $PQ$ . The distance of the centroid of  $\triangle PQR$  from the point of intersection of the line**

$$\frac{x-2}{0} = \frac{y}{2} = \frac{z+3}{-1} \quad \text{and} \quad \frac{x-1}{1} = \frac{y+3}{-3} = \frac{z+1}{1}$$

**is:**

- (1) 69
  - (2) 9
  - (3)  $\sqrt{69}$
  - (4)  $\sqrt{99}$
- 

**17. Let  $R$  be a relation on  $\mathbb{Z} \times \mathbb{Z}$  defined by**

$$(a, b)R(c, d) \text{ if and only if } ad - bc \text{ is divisible by } 5.$$

**Then  $R$  is:**

- (1) Reflexive and symmetric but not transitive
- (2) Reflexive but neither symmetric nor transitive

- (3) Reflexive, symmetric and transitive  
(4) Reflexive and transitive but not symmetric
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**18. If the value of the integral**

$$\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} \left( \frac{x^2 \cos x}{1 + \pi^x} + \frac{1 + \sin^2 x}{1 + e^{\sin x 2023}} \right) dx = \frac{\pi}{4}(\pi + a) - 2,$$

**then the value of  $a$  is:**

- (1) 3  
(2)  $-\frac{3}{2}$   
(3) 2  
(4)  $\frac{3}{2}$
- 

**19. Suppose**

$$f(x) = \frac{(2^x + 2^{-x}) \tan x \sqrt{\tan^{-1}(x^2 - x + 1)}}{(7x^2 + 3x + 1)^3},$$

**then the value of  $f'(0)$  is equal to:**

- (1)  $\pi$   
(2) 0  
(3)  $\sqrt{\pi}$   
(4)  $\frac{\pi}{2}$
- 

**20. Let  $A$  be a square matrix such that  $AA^T = I$ . Then**

$$\frac{1}{2}A \left[ (A + A^T)^2 + (A - A^T)^2 \right]$$

**is equal to:**

- (1)  $A^2 + I$   
(2)  $A^3 + I$   
(3)  $A^2 + A^T$   
(4)  $A^3 + A^T$
-

**21. Equation of two diameters of a circle are  $2x - 3y = 5$  and  $3x - 4y = 7$ . The line joining the points  $(-\frac{22}{7}, -4)$  and  $(\frac{1}{7}, 3)$  intersects the circle at only one point  $P(\alpha, \beta)$ . Then  $17\beta - \alpha$  is equal to:**

- (1) 3
  - (2) 2
  - (3) 4
  - (4) -1
- 

**22. All the letters of the word "GTWENTY" are written in all possible ways with or without meaning, and these words are arranged as in a dictionary. The serial number of the word "GTWENTY" is:**

- (1) 553
  - (2) 531
  - (3) 526
  - (4) 560
- 

**23. Let  $\alpha, \beta$  be the roots of the equation  $x^2 - x + 2 = 0$  with  $\text{Im}(\alpha) > \text{Im}(\beta)$ . Then  $\alpha^6 + \alpha^4 + \beta^4 - 5\alpha^2$  is equal to**

**24. Let  $f(x) = 2^x - x^2, x \in \mathbb{R}$ . If  $m$  and  $n$  are respectively the number of points at which the curves  $y = f(x)$  and  $y = f'(x)$  intersect the x-axis, then the value of  $m + n$  is:**

- (1) 4
  - (2) 5
  - (3) 6
  - (4) 7
- 

**25. If the points of intersection of two distinct conics  $x^2 + y^2 = 4b$  and  $\frac{x^2}{16} + \frac{y^2}{b^2} = 1$  lie on the curve  $y^2 = 3x^2$ , then  $3\sqrt{3}$  times the area of the rectangle formed by the intersection points is ...**

- (1) 432

- (2) 450
- (3) 410
- (4) 415

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**26. If the solution curve  $y = y(x)$  of the differential equation  $(1 + y^2)(1 + \log_e x) dx + x dy = 0, x > 0$ , passes through the point  $(1, 1)$  and**

$$y(e) = \frac{\alpha - \tan\left(\frac{3}{2}\right)}{\beta + \tan\left(\frac{3}{2}\right)},$$

**then  $\alpha + 2\beta$  is:**

- (1) 3
- (2) 4
- (3) 2
- (4) 5

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**27. If the mean and variance of the data 65, 68, 58, 44, 48, 45, 60,  $\alpha, \beta, 60$  where  $\alpha > \beta$  are 56 and 66.2 respectively, then  $\alpha^2 + \beta^2$  is equal to:**

- (1) 6344
- (2) 6500
- (3) 6300
- (4) 6400

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**28. The area (in sq. units) of the part of the circle  $x^2 + y^2 = 169$  which is below the line  $5x - y = 13$  is**

$$\frac{\pi\alpha}{2\beta} \cdot \frac{65}{2} + \frac{\alpha}{\beta} \sin^{-1}\left(\frac{12}{13}\right)$$

**where  $\alpha, \beta$  are coprime numbers. Then  $\alpha + \beta$  is equal to:**

- (1) 171
- (2) 172
- (3) 170
- (4) 168



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**29. If**

$$\binom{11}{1}\binom{11}{2} + \binom{11}{2}\binom{11}{3} + \dots + \binom{11}{9}\binom{11}{10} = \frac{n}{m} \quad \text{with } \gcd(n, m) = 1,$$

**then  $n + m$  is equal to:**

- (1) 2041
- (2) 2035
- (3) 2050
- (4) 2025

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**30. A line with direction ratios  $2, 1, 2$  meets the lines  $x = y + 2 = z$  and  $x + 2 = 2y = 2z$  respectively at the points  $P$  and  $Q$ . If the length of the perpendicular from the point  $(1, 2, 12)$  to the line  $PQ$  is  $l$ , then  $l^2$  is:**

- (1) 65
- (2) 70
- (3) 60
- (4) 68

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**31. In the given circuit, the breakdown voltage of the Zener diode is  $3.0$  V. What is the value of  $I_Z$ ?**

- (1)  $3.3$  mA
- (2)  $5.5$  mA
- (3)  $10$  mA
- (4)  $7$  mA

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**32. The electric current through a wire varies with time as  $I = I_0 + \beta t$ , where  $I_0 = 20$  A and  $\beta = 3$  A/s. The amount of electric charge that crosses through a section of the wire in  $20$  s is:**

- (1)  $80$  C
- (2)  $1000$  C

- (3) 800 C
  - (4) 1600 C
- 

**33. Given below are two statements:**

**Statement I:** If a capillary tube is immersed first in cold water and then in hot water, the height of capillary rise will be smaller in hot water.

**Statement II:** If a capillary tube is immersed first in cold water and then in hot water, the height of capillary rise will be smaller in cold water.

**In the light of the above statements, choose the most appropriate option from the options given below:**

- (1) Both Statement I and Statement II are true
  - (2) Both Statement I and Statement II are false
  - (3) Statement I is true but Statement II is false
  - (4) Statement I is false but Statement II is true
- 

**34. A convex mirror of radius of curvature 30 cm forms an image that is half the size of the object. The object distance is:**

- (1)  $-15$  cm
  - (2) 45 cm
  - (3)  $-45$  cm
  - (4) 15 cm
- 

**35. Two charges of  $5Q$  and  $-2Q$  are situated at the points  $(3a, 0)$  and  $(-5a, 0)$  respectively. The electric flux through a sphere of radius  $4a$  having its center at the origin is:**

- (1)  $\frac{2Q}{\epsilon_0}$
  - (2)  $\frac{5Q}{\epsilon_0}$
  - (3)  $\frac{7Q}{\epsilon_0}$
  - (4)  $\frac{3Q}{\epsilon_0}$
-

**36. A body starts moving from rest with constant acceleration and covers displacement  $S_1$  in the first  $(p - 1)$  seconds and  $S_2$  in the first  $p$  seconds. The displacement  $S_1 + S_2$  will be made in time:**

- (1)  $(2p + 1) \text{ s}$
  - (2)  $\sqrt{2p^2 - 2p + 1} \text{ s}$
  - (3)  $(2p - 1) \text{ s}$
  - (4)  $(2p^2 - 2p + 1) \text{ s}$
- 

**37. The potential energy function (in J) of a particle in a region of space is given as  $U = (2x^2 + 3y^3 + 2z)$ . Here  $x, y,$  and  $z$  are in meters. The magnitude of the  $x$ -component of force (in N) acting on the particle at point  $P(1, 2, 3)$  m is:**

- (1) 2
  - (2) 6
  - (3) 4
  - (4) 8
- 

**38. The resistance  $R = \frac{V}{I}$  where  $V = (200 \pm 5) \text{ V}$  and  $I = (20 \pm 0.2) \text{ A}$ . The percentage error in the measurement of  $R$  is:**

- (1) 3.5%
  - (2) 7%
  - (3) 3%
  - (4) 5.5%
- 

**39. A block of mass 100 kg slides over a distance of 10 m on a horizontal surface. If the coefficient of friction between the surfaces is 0.4, then the work done against friction (in J) is:**

- (1) 4200 J
- (2) 3900 J
- (3) 4000 J
- (4) 4500 J

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**40. Match List I with List II**

List I	List II
A. $\oint \vec{B} \cdot d\vec{l} = \mu_0 i_c + \mu_0 \epsilon_0 \frac{d\phi_E}{dt}$	I. Gauss' law for electricity
B. $\oint \vec{E} \cdot d\vec{l} = -\frac{d\phi_B}{dt}$	II. Gauss' law for magnetism
C. $\oint \vec{E} \cdot d\vec{A} = \frac{Q}{\epsilon_0}$	III. Faraday law
D. $\oint \vec{B} \cdot d\vec{A} = 0$	IV. Ampere – Maxwell law

Choose the correct answer from the options given below:

1. A-IV, B-I, C-III, D-II
  2. A-II, B-III, C-I, D-IV
  3. A-IV, B-III, C-I, D-II
  4. A-I, B-II, C-III, D-IV
- 

**41. If the radius of curvature of the path of two particles of same mass are in the ratio 3:4, then in order to have constant centripetal force, their velocities will be in the ratio of:**

- (1)  $\sqrt{3} : 2$
  - (2)  $1 : \sqrt{3}$
  - (3)  $\sqrt{3} : 1$
  - (4)  $2 : \sqrt{3}$
- 

**42. A galvanometer having coil resistance  $10 \Omega$  shows a full scale deflection for a current of 3 mA. For it to measure a current of 8 A, the value of the shunt should be:**

- (1)  $3 \times 10^{-3} \Omega$
  - (2)  $4.85 \times 10^{-3} \Omega$
  - (3)  $3.75 \times 10^{-3} \Omega$
  - (4)  $2.75 \times 10^{-3} \Omega$
- 

**43. The de-Broglie wavelength of an electron is the same as that of a photon. If the**

velocity of the electron is 25% of the velocity of light, then the ratio of the K.E. of the electron to the K.E. of the photon will be:

- (1)  $\frac{1}{1}$
- (2)  $\frac{1}{8}$
- (3) 8 : 1
- (4)  $\frac{1}{4}$

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**44. The deflection in a moving coil galvanometer falls from 25 divisions to 5 divisions when a shunt of  $24\ \Omega$  is applied. The resistance of the galvanometer coil will be:**

- (1)  $12\ \Omega$
- (2)  $96\ \Omega$
- (3)  $48\ \Omega$
- (4)  $100\ \Omega$

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**45. A biconvex lens of refractive index 1.5 has a focal length of 20 cm in air. Its focal length when immersed in a liquid of refractive index 1.6 will be:**

- (1)  $-16\ \text{cm}$
- (2)  $-160\ \text{cm}$
- (3)  $+160\ \text{cm}$
- (4)  $+16\ \text{cm}$

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**46. A thermodynamic system is taken from an original state A to an intermediate state B by a linear process as shown in the figure. Its volume is then reduced to the original value from B to C by an isobaric process. The total work done by the gas from A to B and B to C would be:**

- (1) 33800 J
- (2) 2200 J
- (3) 600 J
- (4) 1200 J

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47. At what distance above and below the surface of the earth a body will have the same weight (take radius of earth as  $R$ )?

- (1)  $\sqrt{5}R - R$
- (2)  $\frac{\sqrt{5}R - R}{2}$
- (3)  $\frac{R}{2}$
- (4)  $\frac{R}{2}(\sqrt{5} - 1)$

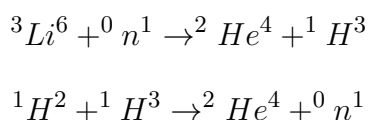
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48. A capacitor of capacitance  $100 \mu\text{F}$  is charged to a potential of  $12 \text{ V}$  and connected to a  $6.4 \text{ mH}$  inductor to produce oscillations. The maximum current in the circuit would be:

- (1)  $3.2 \text{ A}$
- (2)  $1.5 \text{ A}$
- (3)  $2.0 \text{ A}$
- (4)  $1.2 \text{ A}$

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49. The explosive in a Hydrogen bomb is a mixture of  ${}^1\text{H}^2$ ,  ${}^1\text{H}^3$ , and  ${}^3\text{Li}^6$  in some condensed form. The chain reaction is given by



During the explosion, the energy released is approximately

- (1)  $28.12 \text{ MeV}$
- (2)  $12.64 \text{ MeV}$
- (3)  $16.48 \text{ MeV}$
- (4)  $22.22 \text{ MeV}$

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50. Two vessels A and B are of the same size and are at the same temperature. A contains  $1 \text{ g}$  of hydrogen and B contains  $1 \text{ g}$  of oxygen.  $P_A$  and  $P_B$  are the pressures of the gases in A and B respectively, then  $\frac{P_A}{P_B}$  is:

- (1) 16
  - (2) 8
  - (3) 4
  - (4) 32
- 

51. When a hydrogen atom going from  $n = 2$  to  $n = 1$  emits a photon, its recoil speed is  $\frac{x}{5}$  m/s. Where  $x = \underline{\hspace{2cm}}$ . (Use: mass of hydrogen atom =  $1.6 \times 10^{-27}$  kg)

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52. A ball rolls off the top of a stairway with horizontal velocity  $u$ . The steps are 0.1 m high and 0.1 m wide. The minimum velocity  $u$  with which the ball just hits the step 5 of the stairway will be  $\sqrt{x}$  m/s where  $x = \underline{\hspace{2cm}}$  [use  $g = 10$  m/s<sup>2</sup>]

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53. A square loop of side 10 cm and resistance  $0.7 \Omega$  is placed vertically in the east-west plane. A uniform magnetic field of 0.20 T is set up across the plane in the northeast direction. The magnetic field is decreased to zero in 1 s at a steady rate. Then, the magnitude of induced emf is  $\sqrt{x} \times 10^{-3}$  V. The value of  $x$  is         .

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54. A cylinder is rolling down on an inclined plane of inclination  $60^\circ$ . Its acceleration during rolling down will be  $\frac{x}{\sqrt{3}}$  m/s<sup>2</sup>, where  $x = \underline{\hspace{2cm}}$ . (Use  $g = 10$  m/s<sup>2</sup>)

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55. The magnetic potential due to a magnetic dipole at a point on its axis situated at a distance of 20 cm from its center is  $1.5 \times 10^{-5}$  Tm. The magnetic moment of the dipole is

$\text{Am}^2$

Given:  $\frac{\mu_0}{4\pi} = 10^{-7}$  Tm/A

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56. In a double slit experiment shown in figure, when light of wavelength 400 nm is used, a dark fringe is observed at  $P$ . If  $D = 0.2$  m, the minimum distance between the slits  $S_1$  and  $S_2$  is          mm.

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57. A  $16 \Omega$  wire is bent to form a square loop. A 9 V battery with internal resistance  $1 \Omega$  is connected across one of its sides. If a  $4 \mu\text{F}$  capacitor is connected across one of its diagonals, the energy stored by the capacitor will be  $\frac{x}{2} \mu\text{J}$ , where  $x = \underline{\hspace{2cm}}$ .

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58. When the displacement of a simple harmonic oscillator is one third of its amplitude, the ratio of total energy to the kinetic energy is  $\frac{x}{8}$ , where  $x = \underline{\hspace{2cm}}$ .

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59. An electron is moving under the influence of the electric field of a uniformly charged infinite plane sheet  $S$  having surface charge density  $+\sigma$ . The electron at  $t = 0$  is at a distance of 1 m from  $S$  and has a speed of 1 m/s. The maximum value of  $\sigma$  if the electron strikes  $S$  at  $t = 1$  s is  $\alpha \left[ \frac{m\epsilon_0}{e} \right] \text{ C/m}^2$ . The value of  $\alpha$  is

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60. In a test experiment on a model aeroplane in a wind tunnel, the flow speeds on the upper and lower surfaces of the wings are  $70 \text{ ms}^{-1}$  and  $65 \text{ ms}^{-1}$  respectively. If the wing area is  $2 \text{ m}^2$ , the lift of the wing is  $\underline{\hspace{2cm}}$  N.

Given: density of air =  $1.2 \text{ kg m}^{-3}$

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61. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R:

**Assertion A:** The first ionisation enthalpy decreases across a period.

**Reason R:** The increasing nuclear charge outweighs the shielding across the period.

In the light of the above statements, choose the most appropriate from the options given below:

- (1) Both A and R are true and R is the correct explanation of A
  - (2) A is true but R is false
  - (3) A is false but R is true
  - (4) Both A and R are true but R is NOT the correct explanation of A
- 

62. Match List I with List II

LIST-I (Substances)	LIST-II (Element Present)
A. Ziegler catalyst	I. Rhodium
B. Blood Pigment	II. Cobalt
C. Wilkinson catalyst	III. Iron
D. Vitamin B <sub>12</sub>	IV. Titanium



Choose the correct answer from the options given below:

- (1) A-II, B-IV, C-I, D-III
  - (2) A-II, B-III, C-IV, D-I
  - (3) A-III, B-II, C-IV, D-I
  - (4) A-IV, B-III, C-I, D-II
- 

**63. In chromyl chloride test for confirmation of  $\text{Cl}^-$  ion, a yellow solution is obtained. Acidification of the solution and addition of amyl alcohol and 10%  $\text{H}_2\text{O}_2$  turns organic layer blue indicating formation of chromium pentoxide. The oxidation state of chromium in that is**

- (1) +6
  - (2) +5
  - (3) +10
  - (4) +3
- 

**64. The difference in energy between the actual structure and the lowest energy resonance structure for the given compound is**

- (1) electromeric energy
  - (2) resonance energy
  - (3) ionization energy
  - (4) hyperconjugation energy
- 

**65. Given below are two statements:**

**Statement I:** The electronegativity of group 14 elements from Si to Pb gradually decreases.

**Statement II:** Group 14 contains non-metallic, metallic, as well as metalloid elements.

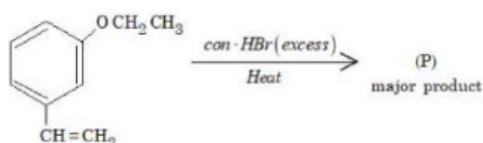
**In the light of the above statements, choose the most appropriate from the options given below:**

- (1) Statement I is false but Statement II is true
- (2) Statement I is true but Statement II is false
- (3) Both Statement I and Statement II are true
- (4) Both Statement I and Statement II are false

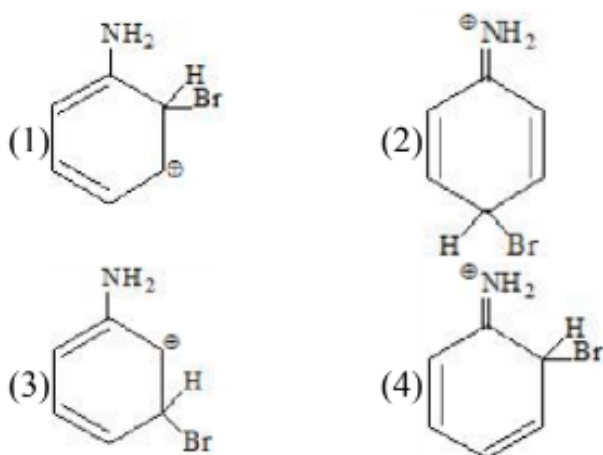
66. The correct set of four quantum numbers for the valence electron of rubidium atom ( $Z = 37$ ) is:

- (1)  $5, 0, 0, +\frac{1}{2}$
- (2)  $5, 0, 1, +\frac{1}{2}$
- (3)  $5, 1, 0, +\frac{1}{2}$
- (4)  $5, 1, 1, +\frac{1}{2}$

67. The major product (P) in the following reaction is:



68. The arenium ion which is not involved in the bromination of Aniline is:



69. Appearance of blood red colour on treatment of the sodium fusion extract of an organic compound with  $\text{FeSO}_4$  in presence of concentrated  $\text{H}_2\text{SO}_4$  indicates the presence of element(s)

- (1) Br
- (2) N
- (3) N and S
- (4) S

70. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R:

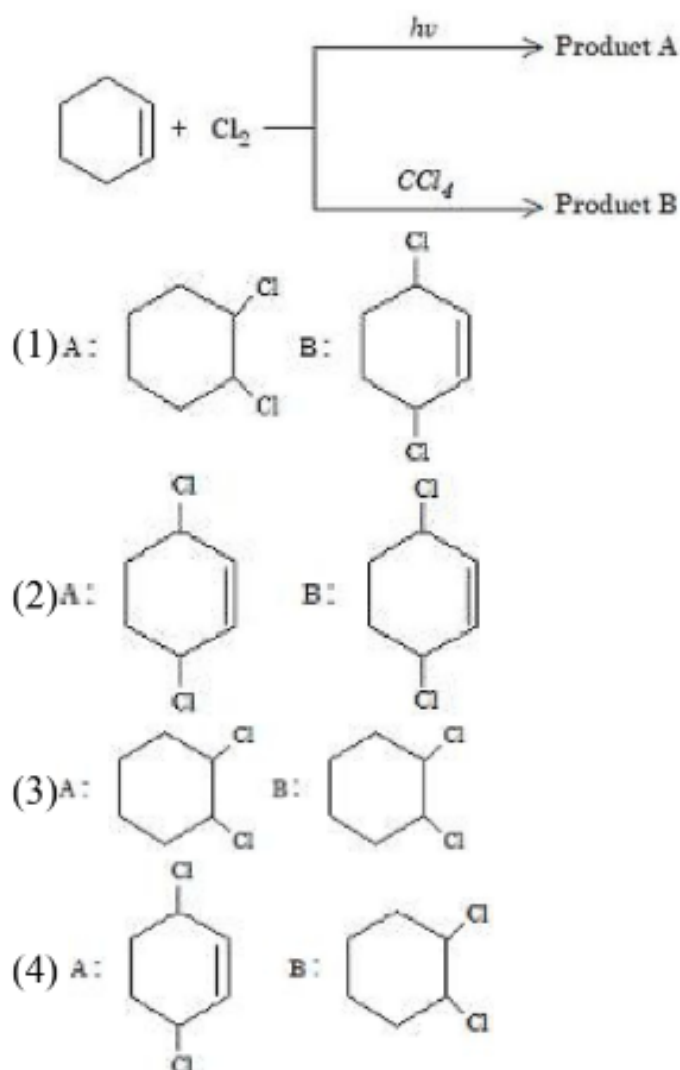
**Assertion A:** Aryl halides cannot be prepared by replacement of hydroxyl group of phenol by halogen atom.

**Reason R:** Phenols react with halogen acids violently.

In the light of the above statements, choose the most appropriate from the options given below:

- (1) Both A and R are true but R is NOT the correct explanation of A
- (2) A is false but R is true
- (3) A is true but R is false
- (4) Both A and R are true and R is the correct explanation of A

71. Identify product A and product B:



---

**72. Identify the incorrect pair from the following:**

- (1) Fluorspar -  $\text{BF}_3$
- (2) Cryolite -  $\text{Na}_3\text{AlF}_6$
- (3) Fluoroapatite -  $3\text{Ca}_3(\text{PO}_4)_2 \cdot \text{CaF}_2$
- (4) Carnallite -  $\text{KCl} \cdot \text{MgCl}_2 \cdot 6\text{H}_2\text{O}$

---

**73. The interaction between  $\pi$  bond and lone pair of electrons present on an adjacent atom is responsible for**

- (1) Hyperconjugation
- (2) Inductive effect
- (3) Electromeric effect
- (4) Resonance effect

---

**74.  $\text{KMnO}_4$  decomposes on heating at 513 K to form  $\text{O}_2$  along with**

- (1)  $\text{MnO}_2$  &  $\text{K}_2\text{O}_2$
- (2)  $\text{K}_2\text{MnO}_4$  &  $\text{Mn}$
- (3)  $\text{Mn}$  &  $\text{KO}_2$
- (4)  $\text{K}_2\text{MnO}_4$  &  $\text{MnO}_2$

---

**75. In which one of the following metal carbonyls, CO forms a bridge between metal atoms?**

- (1)  $[\text{Co}_2(\text{CO})_8]$
- (2)  $[\text{Mn}_2(\text{CO})_{10}]$
- (3)  $[\text{Os}_3(\text{CO})_{12}]$
- (4)  $[\text{Ru}_3(\text{CO})_{12}]$

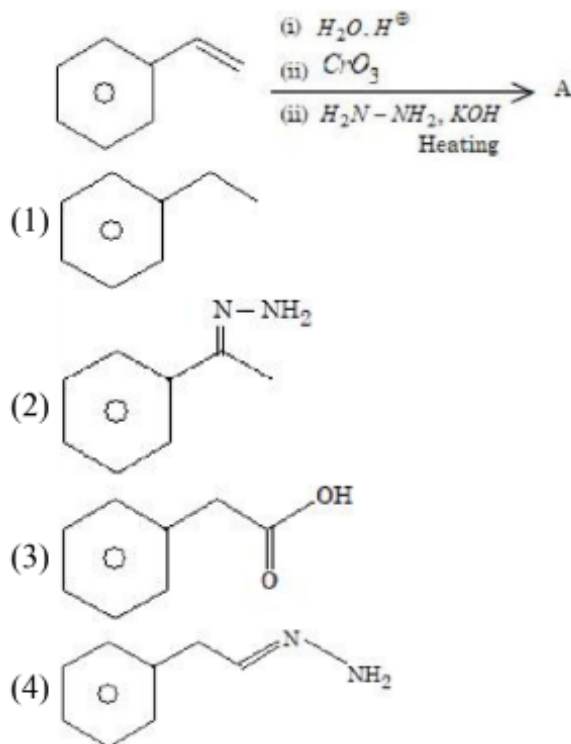
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**76. Type of amino acids obtained by hydrolysis of proteins is:**

- (1)  $\beta$

- (2)  $\alpha$   
 (3)  $\delta$   
 (4)  $\gamma$

77. The final product A formed in the following multistep reaction sequence is:



78. Which of the following is not correct?

- (1)  $\Delta G$  is negative for a spontaneous reaction  
 (2)  $\Delta G$  is positive for a spontaneous reaction  
 (3)  $\Delta G$  is zero for a reversible reaction  
 (4)  $\Delta G$  is positive for a non-spontaneous reaction

79. Chlorine undergoes disproportionation in alkaline medium as shown below:



The values of a, b, c, and d in a balanced redox reaction are respectively:

- (1) 1, 2, 1, and 1  
 (2) 2, 2, 1, and 3

(3) 3, 4, 4, and 2

(4) 2, 4, 1, and 3

---

**80. In alkaline medium,  $\text{MnO}_4^-$  oxidises  $\text{I}^-$  to:**

(1)  $\text{IO}_4^-$

(2)  $\text{IO}^-$

(3)  $\text{I}_2$

(4)  $\text{IO}_3^-$

---

**81. Number of compounds with one lone pair of electrons on central atom amongst following is \_**

$\text{O}_3, \text{H}_2\text{O}, \text{SF}_4, \text{ClF}_3, \text{NH}_3, \text{BrF}_5, \text{XeF}_4$

(1) 2

(2) 3

(3) 5

(4) 4

---

**82. The mass of zinc produced by the electrolysis of zinc sulphate solution with a steady current of 0.015 A for 15 minutes is  $\_ \times 10^{-4}$  g.**

(Atomic mass of zinc = 65.4 amu)

(1) 45.75

(2) 46

(3) 45

(4) 47

---

**83. For a reaction taking place in three steps at the same temperature, the overall rate constant  $K = \frac{K_1 K_2}{K_3}$ . If  $E_{a1}$ ,  $E_{a2}$ , and  $E_{a3}$  are 40, 50, and 60 kJ/mol respectively, the overall  $E_a$  is \_\_\_\_ kJ/mol.**

(1) 20

(2) 40

---

(3) 30

(4) 50

---

84. For the reaction  $\text{N}_2\text{O}_4(g) \rightleftharpoons 2\text{NO}_2(g)$ ,  $K_p = 0.492 \text{ atm}$  at 300K.  $K_c$  for the reaction at the same temperature is  $\dots \times 10^{-2}$ . (Given:  $R = 0.082 \text{ L atm mol}^{-1}\text{K}^{-1}$ )

(1) 1

(2) 2

(3) 3

(4) 4

---

85. A solution of  $\text{H}_2\text{SO}_4$  is 31.4%  $\text{H}_2\text{SO}_4$  by mass and has a density of 1.25 g/mL. The molarity of the  $\text{H}_2\text{SO}_4$  solution is  $\dots \text{ M}$  (nearest integer). (Given molar mass of  $\text{H}_2\text{SO}_4 = 98 \text{ g mol}^{-1}$ )

(1) 3

(2) 5

(3) 6

(4) 4

---

86. The osmotic pressure of a dilute solution is  $7 \times 10^5 \text{ Pa}$  at 273 K. Osmotic pressure of the same solution at 283 K is  $\dots \times 10^4 \text{ Nm}^{-2}$ .

Answer Options:

(1) 72.56

(2) 73

(3) Approximate

(4) None of these

---

87. Number of compounds among the following which contain sulfur as a heteroatom is  $\dots$ .

Compounds: Furan, Thiophene, Pyridine, Pyrrole, Cysteine, Tyrosine

- (1) 2
  - (2) 3
  - (3) None of these
  - (4) None of these
- 

88. The number of species from the following which are paramagnetic and with bond order equal to one is ----.

Species:  $H_2, He_2^+, O_2^+, N_2^-, O_2^{2-}, F_2, Ne_2^+, B_2$

- (1) 1
  - (2) 2
  - (3) More
  - (4) None of these
- 

89. From the compounds given below, the number of compounds which give a positive Fehling's test is ----.

Compounds: Benzaldehyde, Acetaldehyde, Acetone, Acetophenone, Methanal, 4-nitrobenzaldehyde, cyclohexane carbaldehyde.

- (1) 3
  - (2) None of these
  - (3) None
  - (4) All of them
- 

90. Consider the given reaction. The total number of oxygen atoms present per molecule of the product (P) is ----.



- (1) 1
  - (2) Approximate
  - (3) None
  - (4) All
-



## JEE Main 29 Jan Shift 1 Answer Key

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Question Number	Answer
1	(4)
2	(2)
3	(3)
4	(3)
5	(2)
6	(3)
7	(1)
8	(1)
9	(2)
10	(4)
11	(3)
12	(1)
13	(2)
14	(4)
15	(2)
16	(3)
17	(1)
18	(1)
19	(3)
20	(4)
21	(2)

22	553
23	13
24	5
25	432
26	3
27	6344
28	171
29	2041
30	65
31	(2) 5.5 mA
32	(2) 1000 C
33	(3)
34	(1) -15 cm
35	(2)
36	(2)
37	(3)
38	(1) 3.5%
39	(3) 4000 J
40	(3)
41	(1) $\sqrt{3} : 2$
42	(3) $3.75 \times 10^{-3} \Omega$
43	(2) 1/8
44	(2) 96 $\Omega$
45	(2) -160 cm
46	BONUS

47	(4)
48	(2) 1.5 A
49	(4) 22.22 MeV
50	(1) 16
51	(17)
52	(2) $x = 2$
53	(2) $x = 2$
54	(10)
55	(6)
56	(0.20 mm)
57	(81)
58	(9)
59	(8)
60	(810 N)
61	(3)
62	(4)
63	(1)
64	(2)
65	(1)
66	(1)
67	(4)
68	(3)
69	(3)
70	(3)
71	(4)

72	(1)
73	(4)
74	(4)
75	(1)
76	(2)
77	(1)
78	(2)
79	(1)
80	(4)
81	(4)
82	(45.75) or (46)
83	(30)
84	(2)
85	(4)
86	(72.56) or (73)
87	(2)
88	(1)
89	(3)
90	(1)