

JEE Main 2025 April 4 Shift 2 Question Paper

Time Allowed :3 Hour	Maximum Marks :300	Total Questions :75
----------------------	--------------------	---------------------

General Instructions

Read the following instructions very carefully and strictly follow them:

(A) The test is of 3 hours duration. (B) The question paper consists of 75 questions.

The maximum marks are 300. (C) There are three parts in the question paper consisting of Physics, Chemistry and Mathematics having 25 questions in each part of equal weightage. (D) Each part (subject) has two sections.

(i) Section-A: This section contains 20 multiple choice questions which have only one correct answer. Each question carries 4 marks for correct answer and -1 mark for wrong answer.

(ii) Section-B: This section contains 5 questions. The answer to each of the questions is a numerical value. Each question carries 4 marks for correct answer and -1 mark for wrong answer. For Section-B, the answer should be rounded off to the nearest integer.

MATHEMATICS

SECTION-A

1. Let $a > 0$. If the function $f(x) = 6x^3 - 45ax^2 + 108a^2x + 1$ attains its local maximum and minimum values at the points x_1 and x_2 respectively such that $x_1x_2 = 54$, then $a + x_1 + x_2$ is equal to:

- (1) 15
- (2) 18
- (3) 24
- (4) 13

2. Let f be a differentiable function on $\mathbb{R}$ such that $f(2) = 4$. Let

$\lim_{x \rightarrow 0} (f(2+x))^{3/x} = e^\alpha$. Then the number of times the curve

$y = 4x^3 - 4x^2 - 4(\alpha - 7)x - \alpha$ meets the x-axis is:

- (1) 2
 - (2) 1
 - (3) 0
 - (4) 3
-

3. The sum of the infinite series $\cot^{-1} \left(\frac{7}{4} \right) + \cot^{-1} \left(\frac{19}{4} \right) + \cot^{-1} \left(\frac{39}{4} \right) + \cot^{-1} \left(\frac{67}{4} \right) + \dots$ is:

- (1) $\frac{\pi}{2} + \tan^{-1} \left(\frac{1}{2} \right)$
 - (2) $\frac{\pi}{2} - \cot^{-1} \left(\frac{1}{2} \right)$
 - (3) $\frac{\pi}{2} + \cot^{-1} \left(\frac{1}{2} \right)$
 - (4) $\frac{\pi}{2} - \tan^{-1} \left(\frac{1}{2} \right)$
-

4. Let $A = \{-3, -2, -1, 0, 1, 2, 3\}$ and R be a relation on A defined by xRy if and only if $2x - y \in \{0, 1\}$. Let l be the number of elements in R . Let m and n be the minimum number of elements required to be added in R to make it reflexive and symmetric relations, respectively. Then $l + m + n$ is equal to:

- (1) 18
 - (2) 17
 - (3) 15
 - (4) 16
-

5. Let the product of $\omega_1 = (8 + i) \sin \theta + (7 + 4i) \cos \theta$ and $\omega_2 = (1 + 8i) \sin \theta + (4 + 7i) \cos \theta$ be $\alpha + i\beta$, where $i = \sqrt{-1}$. Let p and q be the maximum and the minimum values of $\alpha + \beta$ respectively.

- (1) 140
- (2) 130

(3) 160

(4) 150

6. Let the values of p , for which the shortest distance between the lines $\frac{x+1}{3} = \frac{y}{4} = \frac{z}{5}$ and $\vec{r} = (p\hat{i} + 2\hat{j} + \hat{k}) + \lambda(2\hat{i} + 3\hat{j} + 4\hat{k})$ is $\frac{1}{\sqrt{6}}$, be a, b , where $a < b$. Then the length of the latus rectum of the ellipse $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ is:

(1) 9

(2) $\frac{3}{2}$

(3) $\frac{2}{3}$

(4) 18

7. The axis of a parabola is the line $y = x$ and its vertex and focus are in the first quadrant at distances $\sqrt{2}$ and $2\sqrt{2}$ units from the origin, respectively. If the point $(1, k)$ lies on the parabola, then a possible value of k is:

(1) 4

(2) 9

(3) 3

(4) 8

8. Let the domains of the functions $f(x) = \log_4 \log_3 \log_7(8 - \log_2(x^2 + 4x + 5))$ and $g(x) = \sin^{-1}\left(\frac{7x+10}{x-2}\right)$ be (α, β) and $[\gamma, \delta]$, respectively. Then $\alpha^2 + \beta^2 + \gamma^2 + \delta^2$ is equal to:

(1) 15

(2) 13

(3) 16

(4) 14

9. A line passing through the point $A(-2, 0)$, touches the parabola $P : y^2 = x - 2$ at the point B in the first quadrant. The area of the region bounded by the line AB , parabola P , and the x-axis is:

- (1) $\frac{7}{3}$
 - (2) 2
 - (3) $\frac{8}{3}$
 - (4) 3
-

10. Let the sum of the focal distances of the point $P(4, 3)$ on the hyperbola $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ be $8\sqrt{\frac{5}{3}}$. If for H , the length of the latus rectum is ℓ and the product of the focal distances of the point P is m , then $9\ell^2 + 6m$ is equal to:

- (1) 184
 - (2) 186
 - (3) 185
 - (4) 187
-

11. Let the matrix $A = \begin{pmatrix} 1 & 0 & 0 \\ 1 & 0 & 1 \\ 0 & 1 & 0 \end{pmatrix}$ satisfy $A^n = A^{n-2} + A^2 - I$ for $n \geq 3$. Then the sum of all the elements of A^{50} is:

- (1) 53
 - (2) 52
 - (3) 39
 - (4) 44
-

12. If the sum of the first 20 terms of the series

$$\frac{4.1}{4 + 3.1^2 + 1^4} + \frac{4.2}{4 + 3.2^2 + 2^4} + \frac{4.3}{4 + 3.3^2 + 3^4} + \frac{4.4}{4 + 3.4^2 + 4^4} + \dots$$

is $\frac{m}{n}$, where m and n are coprime, then $m + n$ is equal to:

- (1) 423
- (2) 420

(3) 421

(4) 422

13. If

$$1^2 \cdot \binom{15}{1} + 2^2 \cdot \binom{15}{2} + 3^2 \cdot \binom{15}{3} + \dots + 15^2 \cdot \binom{15}{15} =$$

$2^m 3^n 5^k$, where $m, n, k \in \mathbb{N}$, then $m + n + k$ is equal to: (1)

(1) 19

(2) 21

(3) 18

(4) 20

14. Let for two distinct values of p , the lines $y = x + p$ touch the ellipse $E : \frac{x^2}{4} + \frac{y^2}{9} = 1$ at the points A and B . Let the line $y = x$ intersect E at the points C and D . Then the area of the quadrilateral $ABCD$ is equal to:

(1) 36

(2) 24

(3) 48

(4) 20

15. Consider two sets A and B , each containing three numbers in A.P. Let the sum and the product of the elements of A be 36 and p , respectively, and the sum and the product of the elements of B be 36 and q , respectively. Let d and D be the common differences of A.P.'s in A and B , respectively, such that $D = d + 3$, $d > 0$. If $\frac{p+q}{p-q} = \frac{19}{5}$, then $p - q$ is equal to:

(1) 600

(2) 450

(3) 630

(4) 540

16. If a curve $y = y(x)$ passes through the point $(1, \frac{\pi}{2})$ and satisfies the differential equation

$$(7x^4 \cot y - e^x \csc y) \frac{dx}{dy} = x^5, \quad x \geq 1, \text{ then at } x = 2, \text{ the value of } \cos y \text{ is:}$$

- (1) $\frac{e^2}{64}$
 - (2) $\frac{e^2}{128}$
 - (3) $\frac{e^2}{128} - 1$
 - (4) $\frac{e^2}{64} + 1$
-

17. The center of a circle C is at the center of the ellipse $E : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$, where $a > b$. Let C pass through the foci F_1 and F_2 of E such that the circle C and the ellipse E intersect at four points. Let P be one of these four points. If the area of the triangle PF_1F_2 is 30 and the length of the major axis of E is 17, then the distance between the foci of E is:

- (1) 8
 - (2) 10
 - (3) 12
 - (4) 14
-

18. Let $f(x) + 2f\left(\frac{1}{x}\right) = x^2 + 5$ and

$$2g(x) - 3g\left(\frac{1}{x}\right) = x, \quad x > 0. \text{ If } \alpha = \int_1^2 f(x) dx, \beta = \int_1^2 g(x) dx, \text{ then the value of } 9\alpha + \beta \text{ is:}$$

- (1) 1
 - (2) 0
 - (3) 10
 - (4) 11
-

19. Let A be the point of intersection of the lines

$$L_1 : \frac{x-7}{1} = \frac{y-5}{0} = \frac{z-3}{-1} \quad \text{and} \quad L_2 : \frac{x-1}{3} = \frac{y+3}{4} = \frac{z+7}{5}$$

Let B and C be the points on the lines L_1 and L_2 , respectively, such that

$AB - AC = \sqrt{15}$. Then the square of the area of the triangle ABC is:

- (1) 54
 - (2) 63
 - (3) 57
 - (4) 60
-

20. Let the mean and the standard deviation of the observations 2, 3, 4, 5, 7, a , b be 4 and $\sqrt{2}$ respectively. Then the mean deviation about the mode of these observations is:

- (1) 1
 - (2) $\frac{3}{4}$
 - (3) 2
 - (4) $\frac{1}{2}$
-

SECTION-B

21. If α is a root of the equation $x^2 + x + 1 = 0$ and

$$\sum_{k=1}^n \left(\alpha^k + \frac{1}{\alpha^k} \right)^2 = 20, \quad \text{then } n \text{ is equal to } \text{-----}$$

- (1) 11
 - (2) 10
 - (3) 9
 - (4) 8
-

22. If

$$\int \frac{(\sqrt{1+x^2} + x)^{10}}{(\sqrt{1+x^2} - x)^9} dx = \frac{1}{m} \left(\left(\sqrt{1+x^2} + x \right)^n \left(n\sqrt{1+x^2} - x \right) \right) + C,$$

where $m, n \in \mathbb{N}$ and C is the constant of integration, then $m + n$ is equal to:

- (1) 379
 - (2) 380
 - (3) 381
 - (4) 378
-

23. A card from a pack of 52 cards is lost. From the remaining 51 cards, n cards are drawn and are found to be spades. If the probability of the lost card to be a spade is

$\frac{11}{50}$, then n is equal to -----

- (1) 1
 - (2) 2
 - (3) 3
 - (4) 4
-

24. Let m and n , $m < n$ be two 2-digit numbers. Then the total number of pairs (m, n) such that $\gcd(m, n) = 6$, is -----

- (1) 64
 - (2) 60
 - (3) 50
 - (4) 55
-

25. Let the three sides of a triangle ABC be given by the vectors

$$2\hat{i} - \hat{j} + \hat{k}, \quad \hat{i} - 3\hat{j} - 5\hat{k}, \quad \text{and} \quad 3\hat{i} - 4\hat{j} - 4\hat{k}.$$

Let G be the centroid of the triangle ABC. Then

$6 (|\vec{AG}|^2 + |\vec{BG}|^2 + |\vec{CG}|^2)$ is equal to -----

- (1) 164

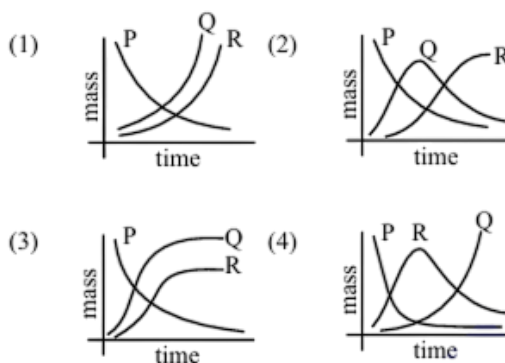
- (2) 166
(3) 162
(4) 160
-

PHYSICS

SECTION-A

26.

A radioactive material P first decays into Q and then Q decays to non-radioactive material R. Which of the following figure represents time dependent mass of P, Q and R?



27.

There are 'n' number of identical electric bulbs, each is designed to draw a power p independently from the mains supply. They are now joined in series across the main supply. The total power drawn by the combination is:

- (1) np
(2) $\frac{p}{n^2}$
(3) $\frac{p}{n}$
(4) p

28.

Consider a rectangular sheet of solid material of length $\ell = 9$ cm and width $d = 4$ cm. The coefficient of linear expansion is $\alpha = 3.1 \times 10^{-5} \text{ K}^{-1}$ at room temperature and one atmospheric pressure. The mass of the sheet is $m = 0.1$ kg and the specific heat capacity $C_v = 900 \text{ J kg}^{-1}\text{K}^{-1}$. If the amount of heat supplied to the material is $8.1 \times 10^2 \text{ J}$, then the change in area of the rectangular sheet is:

- (1) $2.0 \times 10^{-6} \text{ m}^2$
 - (2) $3.0 \times 10^{-7} \text{ m}^2$
 - (3) $6.0 \times 10^{-7} \text{ m}^2$
 - (4) $4.0 \times 10^{-7} \text{ m}^2$
-

29.

Given below are two statements:

Statement (I) : The dimensions of Planck's constant and angular momentum are same.

Statement (II) : In Bohr's model, electron revolves around the nucleus in those orbits for which angular momentum is an integral multiple of Planck's constant.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Both Statement I and Statement II are correct
 - (2) Statement I is incorrect but Statement II is correct
 - (3) Statement I is correct but Statement II is incorrect
 - (4) Both Statement I and Statement II are incorrect
-

30.

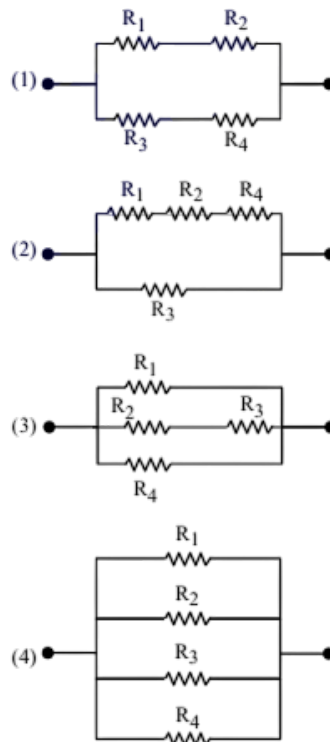
A cylindrical rod of length 1 m and radius 4 cm is mounted vertically. It is subjected to a shear force of 10^5 N at the top. Considering infinitesimally small displacement in the

upper edge, the angular displacement θ of the rod axis from its original position would be: (shear moduli $G = 10^{10} \text{ N/m}^2$)

- (1) $\frac{1}{160\pi}$
 - (2) $\frac{1}{4\pi}$
 - (3) $\frac{1}{40\pi}$
 - (4) $\frac{1}{2\pi}$
-

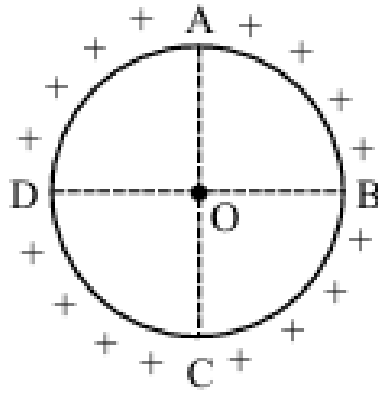
31.

From the combination of resistors with resistance values $R_1 = R_2 = R_3 = 5 \Omega$ and $R_4 = 10 \Omega$, which of the following combination is the best circuit to get an equivalent resistance of 6Ω ?



32.

A metallic ring is uniformly charged as shown in the figure. AC and BD are two mutually perpendicular diameters. Electric field due to arc AB to O is 'E' magnitude. What would be the magnitude of electric field at 'O' due to arc ABC?



- (1) $2E$
 - (2) $\sqrt{2}E$
 - (3) $E/2$
 - (4) Zero
-

33.

There are two vessels filled with an ideal gas where volume of one is double the volume of the other. The large vessel contains the gas at 8 kPa at 1000 K while the smaller vessel contains the gas at 7 kPa at 500 K. If the vessels are connected to each other by a thin tube allowing the gas to flow and the temperature of both vessels is maintained at 600 K, at steady state the pressure in the vessels will be (in kPa).

- (1) 4.4
 - (2) 6
 - (3) 24
 - (4) 18
-

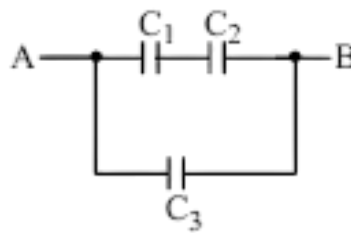
34.

An object is kept at rest at a distance of $3R$ above the earth's surface where R is earth's radius. The minimum speed with which it must be projected so that it does not return to earth is: (Assume M = mass of earth, G = Universal gravitational constant)

- (1) $\sqrt{\frac{GM}{2R}}$
 (2) $\sqrt{\frac{GM}{R}}$
 (3) $\sqrt{\frac{3GM}{R}}$
 (4) $\sqrt{\frac{2GM}{R}}$
-

35.

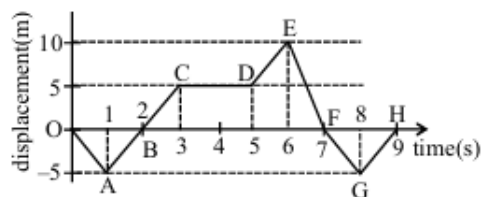
Three parallel plate capacitors C_1 , C_2 , and C_3 each of capacitance $5 \mu\text{F}$ are connected as shown in the figure. The effective capacitance between points A and B, when the space between the parallel plates of C_1 capacitor is filled with a dielectric medium having dielectric constant of 4, is:



- (1) $22.5 \mu\text{F}$
 (2) $7.5 \mu\text{F}$
 (3) $9 \mu\text{F}$
 (4) $30 \mu\text{F}$
-

36.

The displacement x versus time graph is shown below.



The displacement x is plotted against time t . Choose the correct answer from the options given below:

- (A) The average velocity during 0 to 3 s is 10 m/s
- (B) The average velocity during 3 to 5 s is 0 m/s
- (C) The instantaneous velocity at $t = 2$ s is 5 m/s
- (D) The average velocity during 5 to 7 s is the same as instantaneous velocity at $t = 6.5$ s
- (E) The average velocity from $t = 0$ to $t = 9$ s is zero

Choose the correct answer from the options given below:

- (1) (A), (D), (E) only
 - (2) (B), (C), (D) only
 - (3) (B), (D), (E) only
 - (4) (B), (C), (E) only
-

37.

A wheel is rolling on a plane surface. The speed of a particle on the highest point of the rim is 8 m/s. The speed of the particle on the rim of the wheel at the same level as the center of the wheel, will be:

- (1) $4\sqrt{2}$ m/s
 - (2) 8 m/s
 - (3) 4 m/s
 - (4) $8\sqrt{2}$ m/s
-

38.

For the determination of refractive index of glass slab, a travelling microscope is used whose main scale contains 300 equal divisions equals to 15 cm. The vernier scale attached to the microscope has 25 divisions equals to 24 divisions of main scale. The least count (LC) of the travelling microscope is (in cm):

- (1) 0.001

- (2) 0.002
 - (3) 0.0005
 - (4) 0.0025
-

39.

A block of mass 25 kg is pulled along a horizontal surface by a force at an angle 45° with the horizontal. The friction coefficient between the block and the surface is 0.25. The displacement of 5 m of the block is:

- (1) 970 J
 - (2) 735 J
 - (3) 245 J
 - (4) 490 J
-

40.

Two polarisers P_1 and P_2 are placed in such a way that the intensity of the transmitted light will be zero. A third polariser P_3 is inserted in between P_1 and P_2 , at the particular angle between P_1 and P_2 . The transmitted intensity of the light passing the through all three polarisers is maximum. The angle between the polarisers P_2 and P_3 is:

- (1) $\frac{\pi}{4}$
 - (2) $\frac{\pi}{6}$
 - (3) $\frac{\pi}{8}$
 - (4) $\frac{\pi}{3}$
-

41.

Consider a n-type semiconductor in which n_e and n_h are the number of electrons and holes, respectively.

- (A) Holes are minority carriers

- (B) The dopant is a pentavalent atom
 (C) $n_e n_h = n_i^2$ for intrinsic semiconductor
 (D) $n_e \gg n_h$ for extrinsic semiconductor

The correct answer from the options given below is:

- (1) (A) and (C) only
 (2) (B) and (D) only
 (3) (A), (B) and (C) only
 (4) (A), (C) and (D) only
-

42.

Match List-I with List-II.

- | | |
|----------------|--|
| (A) Isobaric | (I) $\Delta Q = \Delta W$ |
| (B) Isochoric | (II) $\Delta Q = \Delta U$ |
| (C) Adiabatic | (III) $\Delta Q = 0$ |
| (D) Isothermal | (IV) $\Delta Q = \Delta U + P\Delta V$ |

Choose the correct answer from the options given below:

- (1) (A)-(IV), (B)-(III), (C)-(II), (D)-(I)
 (2) (A)-(IV), (B)-(I), (C)-(III), (D)-(II)
 (3) (A)-(IV), (B)-(II), (C)-(III), (D)-(I)
 (4) (A)-(I), (B)-(II), (C)-(IV), (D)-(III)
-

43.

Displacement of a wave is expressed as

$$x(t) = 5 \cos \left(628t + \frac{\pi}{2} \right) \text{ m.}$$

The wavelength of the wave when its velocity is 300 m/s is:

- (1) 5 m
 (2) 0.5 m

(3) 0.33 m

(4) 0.33 m

44.

A finite size object is placed normal to the principal axis at a distance of 30 cm from a convex mirror of focal length 30 cm. A plane mirror is now placed in such a way that the image produced by both the mirrors coincide with each other. The distance between the two mirrors is:

(1) 45 cm

(2) 7.5 cm

(3) 22.5 cm

(4) 15 cm

45.

In an electromagnetic system, a quantity defined as the ratio of electric dipole moment and magnetic dipole moment has dimensions of $[ML^2T^{-3}A^{-1}]$. The value of P and Q are:

(1) 1, 0

(2) 1, -1

(3) 1, 1

(4) 0, -1

SECTION-B

46.

A particle of charge $1.6 \mu\text{C}$ and mass $16 \mu\text{g}$ is present in a strong magnetic field of 6.28 T . The particle is then fired perpendicular to magnetic field. The time required for the particle to return to original location for the first time is _____ s. (Take $\pi = 3.14$)

47.

A solid sphere with uniform density and radius R is rotating initially with constant angular velocity (ω_1) about its diameter. After some time during the rotation, it starts losing mass at a uniform rate, with no change in its shape. The angular velocity of the sphere when its radius becomes $\frac{R}{2}$ is ω_2 . The value of x is _____.

48.

If an optical medium possesses a relative permeability of $\frac{10}{\pi}$ and relative permittivity of $\frac{1}{0.0885}$, then the velocity of light is greater in vacuum than in that medium by _____ times.

$$(\mu_0 = 4\pi \times 10^{-7} \text{ H/m}, \quad \epsilon_0 = 8.85 \times 10^{-12} \text{ F/m}, \quad c = 3 \times 10^8 \text{ m/s})$$

49.

In a Young's double slit experiment, two slits are located 1.5 m apart. The distance of screen from slits is 2 m and the wavelength of the source is 400 nm . If the 20 maxima of the double slit pattern are contained within the centre maximum of the single slit diffraction pattern, then the width of each slit is $x \times 10^{-3} \text{ cm}$, where x-value is:

50.

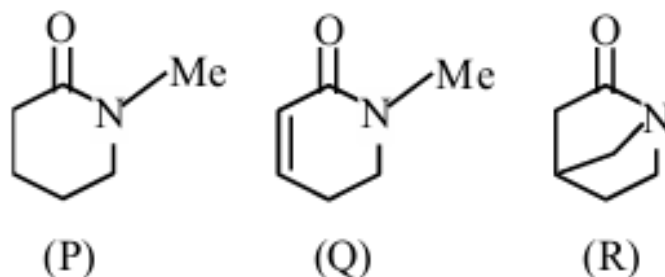
An inductor of self inductance 1 H connected in series with a resistor of 100Ω and an AC supply of 10 V , 50 Hz . Maximum current flowing in the circuit is:

CHEMISTRY

SECTION-A

51.

The correct order of basicity for the following molecules is:



- (1) $P > Q > R$
 - (2) $R > P > Q$
 - (3) $Q > P > R$
 - (4) $R > Q > P$
-

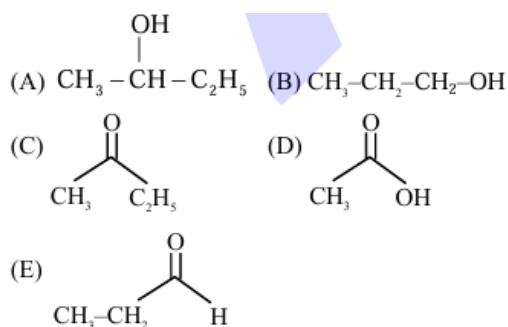
52.

The incorrect relationship in the following pairs in relation to ionisation enthalpies is:

- (1) $Mn^{2+} < Cr^{3+}$
 - (2) $Mn^{2+} < Mn^{3+}$
 - (3) $Fe^{2+} < Fe^{3+}$
 - (4) $Fe^{2+} < Fe^{3+}$
-

53.

Which among the following compounds give yellow solid when reacted with NaOI/NaOH?

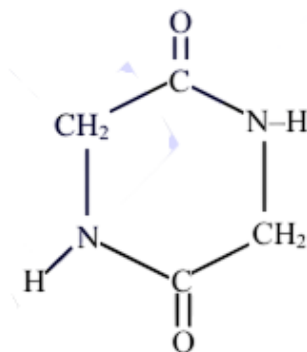


Choose the **correct** answer from the options given below:

- (1) (B), (C) and (E) Only
 - (2) (A) and (C) Only
 - (3) (C) and (D) Only
 - (4) (A), (C) and (D) Only
-

54.

A dipeptide, “x”, on complete hydrolysis gives “y” and “z”; “y” on treatment with aqueous HNO_2 , produces lactic acid. On the other hand, “z” on heating gives the following cyclic molecule.

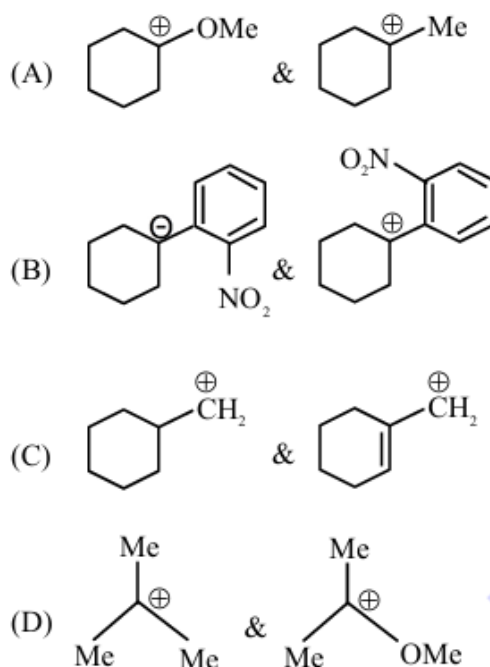


Based on the information given, the dipeptide X is:

- (1) valine-glycine
- (2) alanine-glycine
- (3) valine-leucine
- (4) alanine-alanine

55.

In which pairs, the first ion is more stable than the second?



(1) (B) & (D) only

(2) (A) & (B) only

(3) (B) & (C) only

(4) (A) & (C) only

56.

Given below are two statements:

Statement (I): Alcohols are formed when alkyl chlorides are treated with aqueous potassium hydroxide by elimination reaction.

Statement (II): In alcoholic potassium hydroxide, alkyl chlorides form alkenes by abstracting the hydrogen from the β -carbon.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Both Statement I and Statement II are incorrect
(2) Statement I is incorrect but Statement II is correct
(3) Statement I is correct but Statement II is incorrect
(4) Both Statement I and Statement II are correct
-

57.

Given below are two statements:

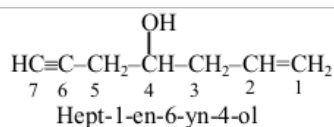
- **Statement (I):** Molal depression constant k_f is given by $\frac{M_1 RT_f}{\Delta S_{\text{fus}}}$, where symbols have their usual meaning.
- **Statement (II):** k_f for benzene is less than the k_f for water.

In light of the above statements, choose the most appropriate answer from the options given below:

- (1) Statement I is incorrect but Statement II is correct
(2) Both Statement I and Statement II are incorrect
(3) Both Statement I and Statement II are correct
(4) Statement I is correct but Statement II is incorrect
-

58.

The IUPAC name of the following compound is:



- (1) 4-Hydroxyhept-1-en-6-yne
(2) 4-Hydroxyhept-6-en-1-yne
(3) Hept-6-en-1-yn-4-ol
(4) Hept-1-en-6-yn-4-ol

59.

Match List-I with List-II:

List-I

List-II

(A) Aniline from aniline-water mixture

(I) Simple distillation

(B) Glycerol from spent-lye in soap industry

(II) Fractional distillation

(C) Different fractions of crude oil in petroleum industry

(III) Distillation at reduced pressure

(D) Chloroform-Aniline mixture

(IV) Steam distillation

Choose the correct answer from the options given below:

(1) (A)-(IV), (B)-(III), (C)-(II), (D)-(I)

(2) (A)-(I), (B)-(II), (C)-(III), (D)-(IV)

(3) (A)-(III), (B)-(II), (C)-(I), (D)-(IV)

(4) (A)-(II), (B)-(I), (C)-(IV), (D)-(III)

60.

A toxic compound “A” when reacted with NaCN in aqueous acidic medium yields an edible cooking component and food preservative “B”. “B” is converted to “C” by dibromane and can be used as an additive to petrol to reduce emission. “C” upon reaction with oleum at 140°C yields an inhalable anesthetic “D”. Identify “A”, “B”, “C”, and “D”, respectively.

(1) Methanol; formaldehyde; methyl chloride; chloroform

(2) Ethanol; acetonitrile; ethylamine; ethylene

(3) Methanol; acetic acid; ethanol; diethyl ether

(4) Acetaldehyde; 2-hydroxyethane; acetic acid; propanoic acid

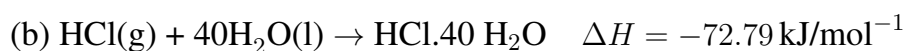
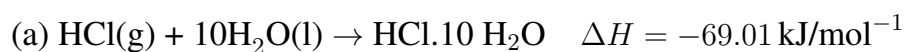
61.

The correct order of $[FeF_6]^{3-}$, $[CoF_6]^{3-}$, $[Ni(CO)_4]$ and $[Ni(CN)_4]^{2-}$ complex species based on the number of unpaired electrons present is:

- (1) $[FeF_6]^{3-} > [CoF_6]^{3-} > [Ni(CN)_4]^{2-} > [Ni(CO)_4]$
 - (2) $[Ni(CN)_4]^{2-} > [FeF_6]^{3-} > [CoF_6]^{3-} > [Ni(CO)_4]$
 - (3) $[CoF_6]^{3-} > [FeF_6]^{3-} > [Ni(CO)_4] > [Ni(CN)_4]^{2-}$
 - (4) $[FeF_6]^{3-} > [CoF_6]^{3-} > [Ni(CN)_4]^{2-} = [Ni(CO)_4]$
-

62.

Consider the given data:



Choose the correct statement:

- (1) Dissolution of gas in water is an endothermic process
 - (2) The heat of solution depends on the amount of solvent
 - (3) The heat of dilution for the HCl (HCl.10H₂O to HCl.40H₂O) is 3.78 kJ/mol
 - (4) The heat of formation of HCl solution is represented by both (a) and (b)
-

63.

Consider the ground state of an atom ($Z = 24$). How many electrons are arranged with Azimuthal quantum number $l = 1$ and $l = 2$ respectively?

- (1) 12 and 4
 - (2) 16 and 4
 - (3) 12 and 5
 - (4) 12 and 5 and 6
-

64.

Given below are two statements:

Statement (I): The first ionisation enthalpy of group 14 elements is higher than the corresponding elements of group 13.

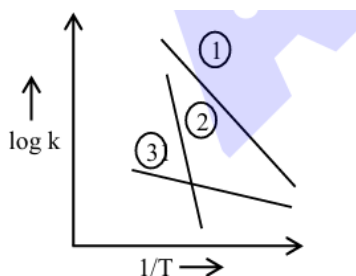
Statement (II): Melting points and boiling points of group 13 elements are in general much higher than those of the corresponding elements of group 14.

Choose the most appropriate answer from the options given below:

- (1) Statement I is correct but Statement II is incorrect
 - (2) Statement I is incorrect but Statement II is correct
 - (3) Both Statement I and Statement II are incorrect
 - (4) Both Statement I and Statement II are correct
-

65.

Consider the following plots of log of rate constant $k(\log k)$ vs $\frac{1}{T}$ for three different reactions. The correct order of activation energies of these reactions is:



Choose the correct answer from the options given below:

- (1) $Ea_2 > Ea_1 > Ea_3$
 - (2) $Ea_1 > Ea_3 > Ea_2$
 - (3) $Ea_1 > Ea_2 > Ea_3$
 - (4) $Ea_3 > Ea_2 > Ea_1$
-

66.

‘X’ is the number of electrons in t_{2g} orbitals of the most stable complex ion among $[Fe(NH_3)_6]^{3+}$, $[Fe(Cl)_6]^{3-}$, $[Fe(C_2O_4)_3]^{3-}$ and $[Fe(H_2O)_6]^{3+}$.

The nature of oxide of vanadium of the type V_2O_x is:

- (1) Acidic
 - (2) Neutral
 - (3) Basic
 - (4) Amphoteric
-

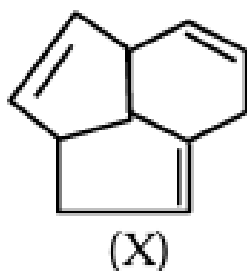
67.

The elements of Group 13 with highest and lowest first ionisation enthalpies are respectively:

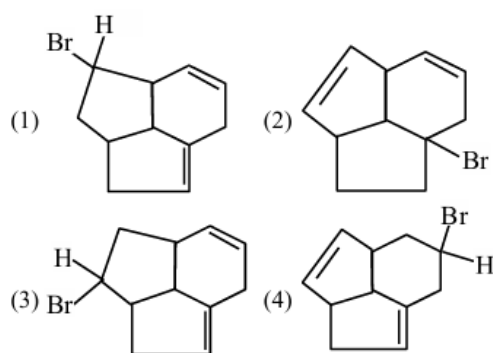
- (1) B & Ga
 - (2) B & Tl
 - (3) Ti & B
 - (4) B & In
-

68.

Consider the following molecule (X).



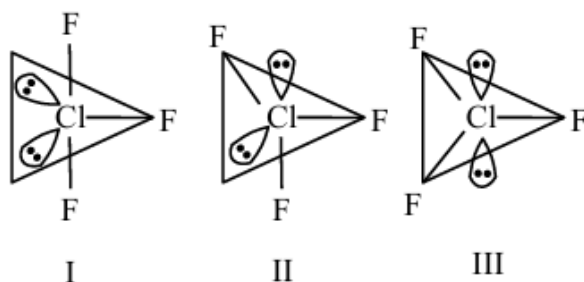
The structure of X is:



69.

Given below are two statements:

Statement (I) : for ClF_3 , all three possible structures may be drawn as follows.



Statement (II): Structure III is most stable, as the orbitals having the lone pairs are axial, where the $\ell p - \beta p$ repulsion is minimum.

In light of the above statements, choose the most appropriate answer from the options given below:

- (1) Statement I is incorrect but Statement II is correct
- (2) Statement I is correct but Statement II is incorrect
- (3) Both Statement I and Statement II are correct
- (4) Both Statement I and Statement II are incorrect

70.

Half-life of zero-order reaction $A \rightarrow \text{product}$ is 1 hour, when initial concentration of reaction is 2.0 mol L^{-1} . The time required to decrease concentration of A from 0.50 to 0.25 mol L^{-1} is:

- (1) 0.5 hour
 - (2) 4 hour
 - (3) 15 min
 - (4) 60 min
-

SECTION-B

71.

Sea water, which can be considered as a 6 molar (6 M) solution of NaCl, has a density of 2 g mL^{-1} . The concentration of dissolved oxygen (O_2) in sea water is 5.8 ppm. Then the concentration of dissolved oxygen (O_2) in sea water, in $x \times 10^{-6} \text{ m}$. $x = \text{.....}$. (Nearest integer)

Given: Molar mass of NaCl is 58.5 g mol^{-1}

Molar mass of O_2 is 32 g mol^{-1} .

72.

The amount of calcium oxide produced on heating 150 kg limestone (75% pure) is kg. (Nearest integer)

Given: Molar mass (in g mol^{-1}) of Ca-40, O-16, C-12

73.

A metal complex with a formula $\text{MCl}_4\text{3NH}_3$ is involved in sp^3d^2 hybridisation. It upon reaction with excess of AgNO_3 solution gives 'x' moles of AgCl. Consider 'x' is equal to

the number of lone pairs of electron present in central atom of BrF_5 . Then the number of geometrical isomers exhibited by the complex is _____

74.

The molar conductance of an infinitely dilute solution of ammonium chloride was found to be $185 \text{ S cm}^{-1} \text{ mol}^{-1}$ and the ionic conductance of hydroxyl and chloride ions are 170 and $70 \text{ S cm}^{-1} \text{ mol}^{-1}$, respectively. If molar conductance of 0.02 M solution of ammonium hydroxide is $85.5 \text{ S cm}^{-1} \text{ mol}^{-1}$, its degree of dissociation is given by $x \times 10^{-1}$. The value of x is _____. (Nearest integer)

75.

x mg of Mg(OH)_2 (molar mass = 58) is required to be dissolved in 1.0 L of water to produce a pH of 10.0 at 298 K . The value of x is ____ mg. (Nearest integer)
(Given: Mg(OH)_2 is assumed to dissociate completely in H_2O)
